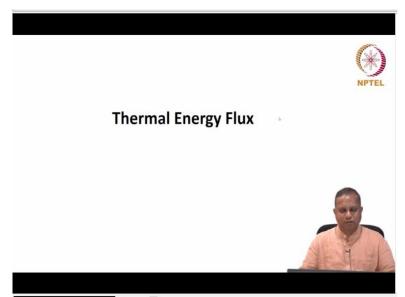
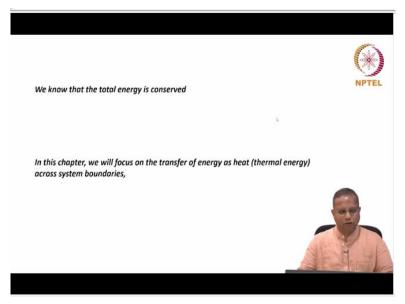
### Transport Phenomena in Biological Systems Prof. G. K. Suraishkumar Department of Biotechnology Bhupat and Jyoti Mehta School of Biosciences Building Indian Institute of Technology-Madras Lecture-80 Course Review-Part 3



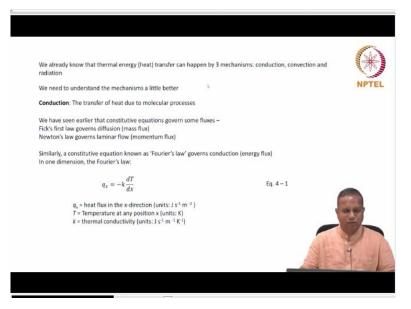
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Welcome to the third review lecture. In this lecture we look at thermal energy flux as well as the charge flux. The thermal energy flux is a thermal heat transport is more relevant to us. However, the total energy is conserved, thermal energy alone is not conserved. Therefore, we looked at total energy and then backed out thermal energy part of it on one side and that is how we could write this expression.

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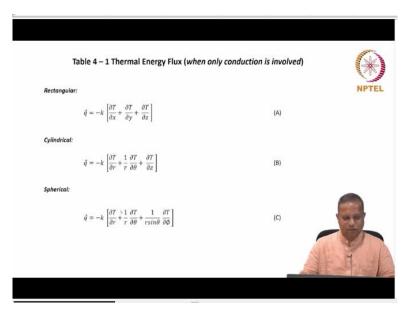


So, we also saw before attempting an equation of heat energy, that heat is transferred by 3 major mechanisms, one is conduction due to molecular processes. The other one is convection due to movement of the medium bulk flow. The third is radiation, which is through electromagnetic waves and so on so forth. It can happen even in free space.

### (Refer Slide Time: 01:22)

 2				
In three dimens	ions, in an isotropic medium, $k \neq f($	x, y, z)		(*)
	$\vec{q} = -k  \vec{\nabla} T$		Eq. 4 – 2	NPTEL
Table 4 – 1 gives	the component-wise equations in the	three coordinate systems		
In a moving fluid	d, $ec{q}$ represents the flux of thermal ener	gy relative to the local velo	city	
Now, let us defin	ne a quantity called thermal diffusivity:	a		
	$\alpha \equiv \frac{k}{\rho \widehat{c_p}}$	ъ	Eq. 4 – 3	
Units of a:	$\frac{j m^{-1} s^{-1} K^{-1}}{kg m^{-3} j kg^{-1} K^{-1}} = m^2 s^{-1}$			
Can you compar	re the units of $\alpha$ (heat energy) with tho	se of D (mass) and $v = \frac{\mu}{\rho}$ (n	nomentum)?	8
What did you fir	nd?			IL

The first we saw the equation for conduction alone the Fourier law  $q = -k\nabla T$  okay dT/dx in 1 dimension or  $\nabla T$  in 3 dimensions, then  $\alpha$ , the thermal diffusivity  $k/\rho C_p$  has the same units as that of mass diffusivity, as well as intrinsic viscosity  $\mu/\rho$  okay. So,  $\mu/\rho$  has same units,m<sup>2</sup>s<sup>-1</sup>. So, they are all equivalent physical quantities is what we can say for different transports. (**Refer Slide Time: 02:05**)



Then this table gives you the thermal energy flux when only one only conduction is involved, you are asked to make a copy of it and keep it as a part of your notes, for these 3 different coordinate systems okay.

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1

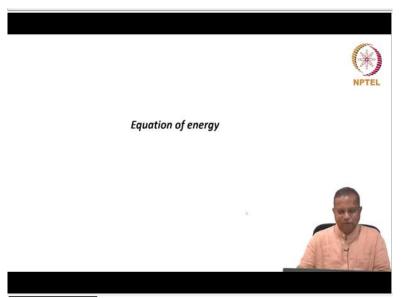
Convection: Flow induced heat flux	
Two kinds of convection exist:	NPTEL
Forced convection: heat transfer due to flow generated by an external means such as a pressure g by a pump or a blower	gradient caused
Free convection: heat transfer due to a flow, normally small in magnitude, which is generated by a which in turn is caused by a heating/cooling	a density differential,
We will see much more of convective heat transport in a later chapter	
6	

(Refer Slide Time: 02:22)

	1
Radiation: Heat transport through electromagnetic waves	
From early physics/chemistry we know that the transitions of electration atom result in emission of radiation	rons between various energy levels in an NI
Thus, any substance at an absolute temperature of T K > 0 K will en Further, when any electromagnetic energy is incident on a substant transitions	
When the energy is transferred as heat through radiation, from say by Stefan-Blotzmann's law:	a body to its surroundings, the radiative flux is given
$q_r = \sigma \epsilon \left( T_{body}^4 - T_{surr}^4 \right)$	Eq. 4.1 1
σ: the Stefan-Boltzmann constant = 5.67 X 1	0.8 W m <sup>-2</sup> K <sup>-4</sup>
c: is emissivity of the body T: the absolute temperature	

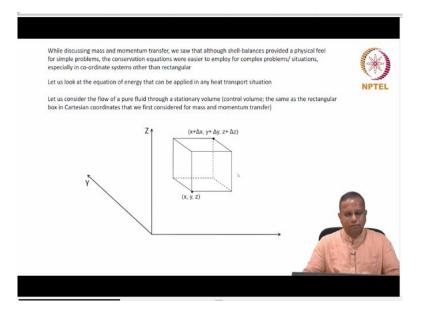
Then we had looked at the equation of energy.

# (Refer Slide Time: 02:33)



We did not spend as much time with the equation of energy.

(Refer Slide Time: 02:38)



However, the equation of energy, the same way, before that let me say this, again that you could do 2 approaches here, one is shell balances, the other one is derive the conservation equation, write it in a form that would be useful and use it directly for more situations. I did not explicitly show you, shell balances here, I went directly to the equation approach.

(Refer Slide Time: 03:05)

Let us consider the relevant energies	()
Internal energy, which can be visualized as arising from the vibrational, rotational and potential energies of the molecule	NPTEL
Kinetic energy, which is associated with the observable (bulk) motion	
<ul> <li>Potential energy (to begin with, it is clubbed with the work done term because it can be interpreted as the work done against gravity)</li> </ul>	
Energy that crosses the control volume boundaries as heat through conduction	
Energy that is generated as heat in the control volume by say, metabolic activities	
Work done against the stresses (and other aspects, such as gravity)	
<ul> <li>Other energies (say electrical, magnetic, surface, etc.,), which we will ignore now – they can be added to the total energy term in the final equation by mere algebraic addition, if needed</li> </ul>	
	1-1-

And that was the outline of how we got.

(Refer Slide Time: 03:09)

Let us write the law of conservation of energy, in our intuitive balance way, as $\frac{d(E)}{d(E)} = (r_{EL} - r_{EO}) + (r_{Ea} - r_{EC})$	()
$\frac{d(E)}{dt} = (r_{Et} - r_{Eo}) + (r_{Eg} - r_{Ec})$ Energy that accumulates IN the system the system boundaries IN the system the system boundaries Energy that CROSSES IN the system the system boundaries IN the system and work components (thermodynamics)	NPTEL
Let us further separate the convection and conduction aspects:	
$ \begin{cases} Rate of accumulation \\ of I.E + K.E \end{cases} $	
$= \begin{cases} Net \ rate \ of \ I, E + K.E \\ \{ ln \ by \ convection \\ by \ generation, say \\ metabolic \\ \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ metabolic \\ \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ metabolic \\ \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \end{cases} + \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} + \end{cases} + \end{cases} + \end{cases} + \begin{cases} (Net \ rate \ of \ heat \ addition \\ by \ generation, say \\ etc., \end{cases} + \end{cases} $	
I.E.: internal energy	-
K.E.: kinetic energy	

There was discussed in some detail.

### (Refer Slide Time: 03:13)

arrive at a useful expression Note that that the total er	various aspects, term by term, u on for <b>thermal</b> energy transport nergy is conserved, but the therm transport is of interest to us in th	nal energy alone is not con		
It is recommended that the	the derivation here, but it is give he learner goes through the deriv			
Here, we will directly pres	sent the equation			
$\frac{\partial}{\partial t} \rho \left( \widehat{U} + \frac{1}{2} v^2 \right)$	$= -\left(\vec{\nabla}.\rho\vec{v}\left(\hat{U}+\frac{1}{2}v^2\right)\right)$	$-(\vec{\nabla}.\vec{q})$	$+\rho(\vec{v}.\vec{g})$	
Rate of energy gain puv	rate of energy in puv by convection	rate of energy in puv by conduction	Rate of work done on the fluid puv by gravitational forces	
	$-(\vec{\nabla}.p\vec{v})$	$-(\vec{\nabla}.[\hat{\tau},\vec{v}]) + \dot{Q}$	other - Wother	
	Rate of work done on the fluid puv by pressure forces		y metabolic	5

They put everything together. We got the equation of energy as  $\frac{\partial}{\partial t}\rho(\hat{U}+1/2v^2)$ . Therefore rate of energy gain per unit volume of the control volume equals -  $\nabla \cdot \rho v(\hat{U}+1/2v^2)$ . This is rate of energy in per unit volume by convection, (- $\nabla \cdot q$ ) rate of energy in per unit volume conduction, (+ $\rho v.g$ ), the rate of work done on the fluid per unit volume by gravitational forces -  $\nabla$ .pv.

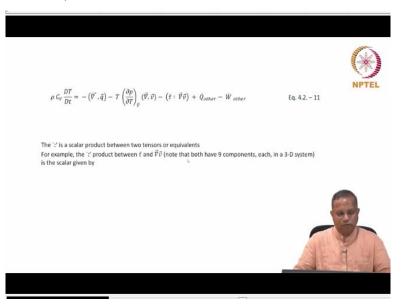
The rate of work done on the fluid per unit volume by pressure forces,  $-\nabla \cdot (\tau \cdot v)$  rate of work done on the fluid per unit volume by viscous forces  $+\dot{Q}$  (other it could be a metabolic)  $-\dot{W}_{other}$  okay, we saw other shop work and so on okay.

In vector notation

$$\begin{aligned} \frac{\partial}{\partial t} \rho \left( \hat{U} + \frac{1}{2} v^2 \right) &= - \left( \vec{\nabla} . \rho \vec{v} \left( \hat{U} + \frac{1}{2} v^2 \right) \right) &- (\vec{\nabla} . \vec{q}) &+ \rho (\vec{v} . \vec{g}) \\ \text{Rate of} & \text{Rate of} & \text{Rate of} & \text{Rate of} & \text{energy in, puv} \\ \text{gain puv} & \text{by convection} & \text{by conduction} & \text{fluid puv by} \\ &- (\vec{\nabla} . p \vec{v}) &- (\vec{\nabla} . [\vec{\tau} . \vec{v}]) \\ \text{Rate of work done} & \text{Rate of work done} & \text{on the fluid puv by} \\ & \text{on the fluid puv by} & \text{on the fluid puv by} \\ & \text{pressure forces} & \text{viscous forces} \\ &+ \dot{Q}_{\text{say, other like metabolic heat}} - \dot{W}_{\text{other}} & (4.2-3) \end{aligned}$$

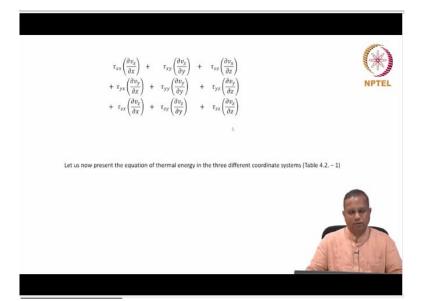
where puv is per unit volume.

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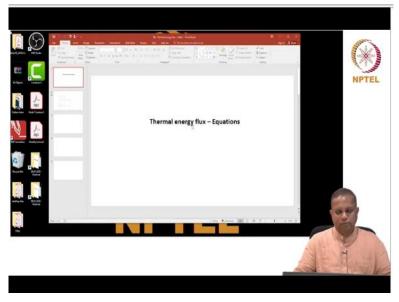
Other work and so on. So, that we wrote in a useful form, we looked at the double dot product which is a scalar product between 2 tensors or equivalents.

## (Refer Slide Time: 04:20)



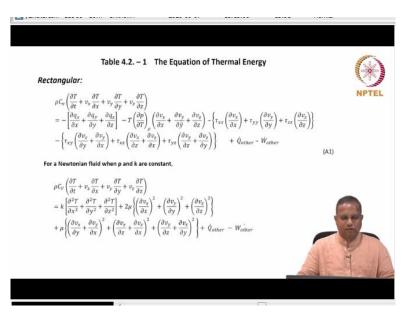
Which also has 9 components.

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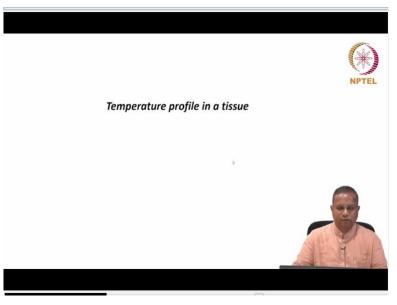
And then I showed you the application of course they were tables of the energy equation, which I would asked you to make a copy off and keep as a part of your notes.

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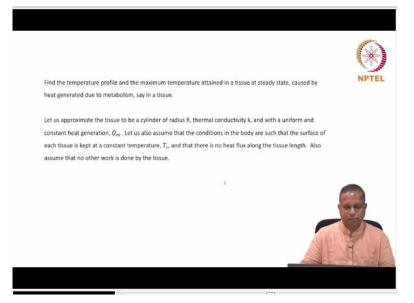
Again, this flashing this here I am not going to discuss this here. Then I showed you how we discussed.

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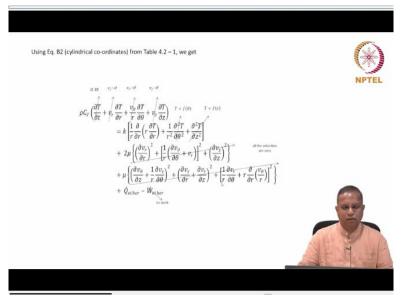


How you could use the equation of thermal energy to get to the temperature profile in a tissue, this was the first example.

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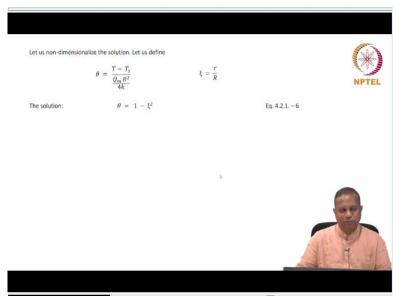


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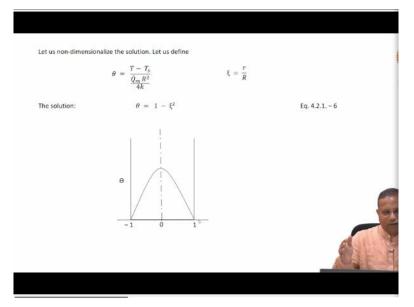
Let us take $\dot{Q}_{other} = \dot{Q}_m$ = metabolic heat rate. Then, we can write		(*)
$-\frac{k}{r}\frac{d}{dr}\left(r\frac{dT}{dr}\right)=\dot{Q}_{m}$	Eq. 4.2.1 2	NPTEL
B.C. 1: at $r = 0$ , $T = finite \text{ or } \frac{dT}{dr} = 0$	Eq. 4.2.1 3	
B.C. 2: at $r = R_c$ $T = T_{\mu}$	Eq. 4.2.1 4	
Integrating Eq. 4.2.1 2 once with B.C. 1, we get		
$\frac{dT}{dr} = -\frac{\dot{Q}_m}{2k}r$		
Integrating again with B.C. 2, we get $^{\rm b}$		-
$T = T_s + \frac{\hat{Q}_m R^2}{4k} \left  1 - \left(\frac{r}{\hat{R}}\right)^2 \right $	Eq. 4.2.1, -5	
		Shark

Then by cancelling the irrelevant terms we could get to the governing equation.

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Which we can solve in a fashion similar to what we did for laminar flow through a pipe. (**Refer Slide Time: 05:19**)



To get a parabolic profile of non dimensional temperature verses non dimensional dimension. (Refer Slide Time: 05:28)

	ation at the cylindrical surface, for the tissue k		
	$= 2\pi RL \times q_r  _{r-R}$	Eq. 4.2.1 7	NPT
	$= 2\pi RL \left(-k \frac{dr}{dr}\right)\Big _{\tau = \pi} = \pi R^2$	L Q̂ <sub>m</sub> Eq. 4.2 .1 8	
From Eq. 4.2.1 5,	mux occurs where r = 0		
Thus,	$T_{max} = T_s + \frac{\dot{Q}_m R^2}{4k}$		
For typical values, sa	y R = 1 cm, $\dot{Q}_{\rm m} = 5$ cal cm $^3$ h $^1$ , k = 10 $^3$ cal (cm	n.s.°C) $^{\circ}$ and $T_{i}$ = 37°C	
	$T_{max} = 37 + \frac{5 \times 1^2}{(4 \times 10^{-3})3600} = 37.3 ^{\circ}\text{C}$		
The temperature at	the centre of the tissue could be $0.3{}^{\rm e}{\rm C}$ higher	than at the surface	-

And also we look at the rate of heat dissipation at the cylindrical surface. Then we plugged in some numbers to see that the maximum temperature is about 0.3 degrees higher or could be about 0.3 degrees higher compared to this surface temperature, ok. Then we saw an example with unsteady state right, this is been the pattern we looked at steady state cases and then unsteady state. Unsteady state just to give you a flavour of the complexities involved and what does unsteady state really mean where all does it become relevant and so on so forth.

The way you should start looking for unsteady state, ok despite the mathematical complexity that is.

#### (Refer Slide Time: 06:12)



And an example was given that complicated with the complications.

# (Refer Slide Time: 06:17)

	A SECTION STREET	TU-second second		a a a a	NPTEL
	Six are dealing with a sphere. Thus, 4 is most converse to our sphe from Eq. (2) of Table 41. – 1, wher sam		n, ve př		a de la contra contr Contra contra c
	$\label{eq:alpha} \begin{split} \frac{dT}{dt} &= \left(\frac{k}{\rho C_0}\right) \frac{1}{r^2} \frac{d}{dr} \bigg( r \\ int us define & \left(\frac{k}{\rho C_0}\right) \ i = a \end{split}$	$\left(\frac{dT}{dr}\right)$	fa.4.1.11	8	
	Let us say that the drop surface temper $\label{eq:LL} LE:=-for 0 < < < 0,$ $BL(1)=-for + 0,$	110 7	<ul> <li>C (2) at the start of the space, 1 + 0</li> <li>T<sub>0</sub></li> <li>(6, 4.2.2, -1</li> <li>(6, 4.2.2, -1)</li> </ul>		
	4		<ul> <li>Fi Bi 432 -4</li> <li>Fi Bi 1 A 1</li> </ul>		25
2				IE	The

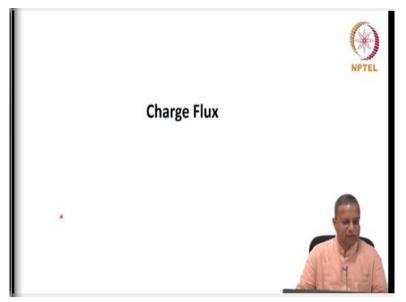
In the mathematical part of it.

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And the it is a very relevant situation, ok, we looked at a micro analysis system the time that it takes for the entire droplet to reach a certain temperature which is necessary for a certain reaction to occur which could be one of the key reactions in the step of processes, that are relevant for that particular system.

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Then we looked at charge flux, ok. We said we biological engineers need to look at charge flux in some detail.

## (Refer Slide Time: 06:57)

The fur	ndamental biomolecules:		(A)
	Lipids	Charged	NPTE
	Carbohydrates	Many are charged	
	Proteins	Charged	
	Nucleic acid	Charged	
and hearing		ty to sense our environment through sight, smell, t g of our nervous system, our brain and our heart	taste, touch
	Charge is a fundamental ph	ysical quantity that is conserved	
		onships related to electrical charges and consequen to clenificantly enuin the maninulator of hinkey	

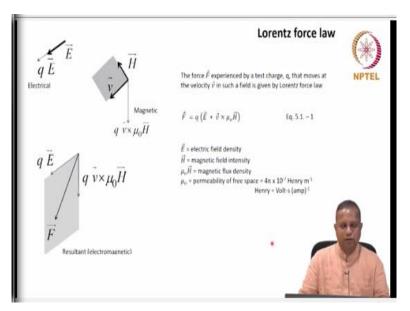
Because the fundamental biomolecules could all be charged, ok, lipids are always charged, carbohydrates are many of them are charged, proteins are charged, nucleic acids are charged. And therefore we are dealing with charged particles has the fundamental units of all our systems and therefore we need to look at charge flux is what was mentioned.

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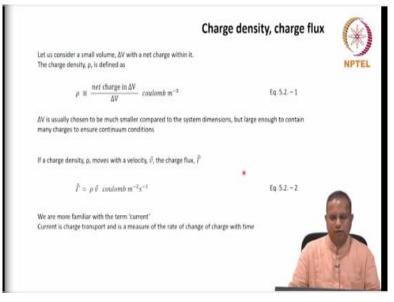
And we looked at some background you know the we derived the charge balance equation in the same way that we derive the other balance equations.

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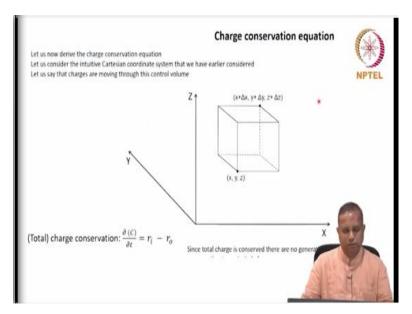
Before that the force that is experience by a particle that is moving in a electromagnetic field is given by Lorentz force law,  $F = q (E + v \times \mu_0 H)$ , this is we reviewed this.

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And then we define charge density, charge flux.

## (Refer Slide Time: 08:04)



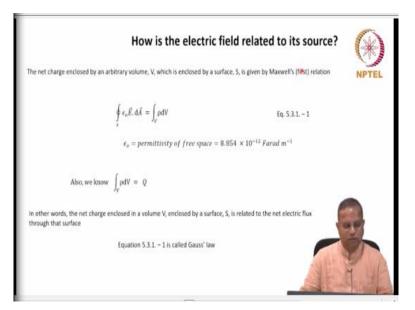
And then based on those quantities we derive the charge conservation equation.

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	he NET charge density	Let I' be the NET charge flux	(*)
	n a sa ann an ann an ann an an ann an ann an	tes of (accumulation = input – output]) can be writh $l_x^{\prime}\Big _x \Delta x \Delta y\Big] - \left[l_x^{\prime}\Big _{x + \Delta x} \Delta y \Delta x + l_y^{\prime}\Big _{y + \Delta y} \Delta x + l_y^{\prime}\Big _{x + \Delta y} \Delta x + l_y^{\prime}\Big _{y + \Delta x} + l_y^{\prime}\Big _{y + \Delta x} \Delta x + l_y^{\prime}\Big _{y + \Delta x} + l_y^{\prime}\Big _{z +$	NPTEL
		$\left[r_{g}\right]_{g} (\Delta x \Delta y) = \left[r_{g}\right]_{g+\Delta x} (\Delta y \Delta x + y)_{y+\Delta y} (\Delta x \Delta x + 0)$ the limits as $\Delta x \rightarrow 0$ , $\Delta y \rightarrow 0$ , $\Delta z \rightarrow 0$	(I) <sub>2+42</sub> (mov)]
	$\frac{\partial \rho}{\partial t} = -\left(\frac{\partial l_x'}{\partial x} + \frac{\partial l_y'}{\partial y} + \frac{\partial l_z'}{\partial z}\right)$		
	$\frac{\partial \rho}{\partial t} + \vec{\nabla} \cdot \vec{l} = 0$	Charge conservation equation	Eq. 5.1.
	- electr	urge conservation, in terms of the relevant effects ic and magnetic fields – by the Maxwell's equations	-
_			A A A A A A A A A A A A A A A A A A A

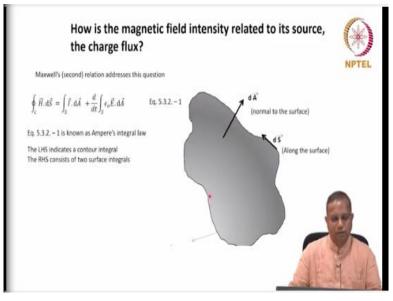
Rather simple derivation compared to the other things and we could get  $\frac{\partial \rho}{\partial t} + \nabla I' = 0$ , the charge flux here equals 0, the differential form of the charge conservation equation. Then you are introduced to the Maxwell's relations, Maxwell's equations which are the fundamental equations for all electromagnetism.

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They answer these questions, first one answers how is the electric field related to it is source, the Gauss' law.

## (Refer Slide Time: 08:42)



Second one answers how is the magnetic field intensity related to it is source, the charge flux, that results in Ampere's integral law.

## (Refer Slide Time: 08:52)

In other words, the line integral (circulation) of the magnetic field is equal to the sum of	d intensity, $\vec{H}_i$ around a closed contour	(*
the net current passing through the sur the time rate of change of the net displa	rface spanning the contour and lacement flux density (displacement current) through the surface	NPTE
Eq 5.3.2. – 1 can be written as		
$\oint \vec{H}.\mathrm{d}\vec{S} = I + \epsilon_0 \frac{d\phi_E}{dt}$	Eq. 5.3.2. – 2	
$\phi_{\mathcal{E}}$ : electric 'flux' (historically called flux, a	nd we just use quotes to avoid confusion in our context)	
I: current		
In other words, an electric current and a time-w	ariant electric 'flux' produce a magnetic field	1
		H

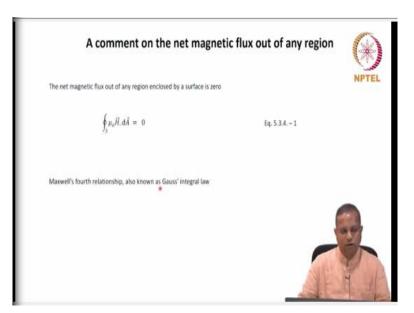
Then the.

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How are electric field an	d magnetic flux related?
$\oint_{c} \vec{E}. d\vec{S} = -\frac{d}{dt} \int_{S} \mu_{a} \vec{H}. d\vec{A}$	tq. 5.3.31
Maxwell's third relationship also known as Faraday's integral law	
In terms of the magnetic 'flux', $\phi_{\rm P},$ this is written as:	1
$\oint \vec{E}.\mathrm{d}S = -\frac{d\phi_{\theta}}{dt}$	Eq. 5.3.3 2

Third Maxwell's equation describes how are electric field and magnetic flux related, ok that is the Faraday's integral law.

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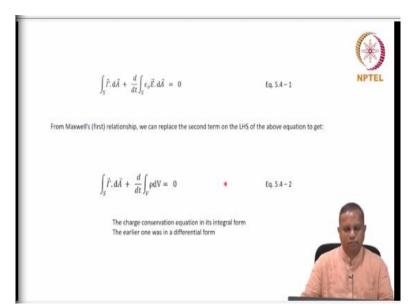
Then a comment on the net magnetic flux out of any region was given by Maxwell's fourth equation which is also called the Gauss' integral law.

# (Refer Slide Time: 09:17)

n of charge ion in the system	NPTEL
charge leaving through the surface bour	adaries of the system
he rate of decrease of charge within the	system
$e_{o}\vec{E}.\mathrm{d}\vec{A}$	
$\epsilon_0 \vec{E}$ , d $\vec{A}$	35
	ion in the system I charge leaving through the surface bour

Then we saw.

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(Refer Slide Time: 09:31)



A few simplifications that make our life easier especially when we deal with electromagnetism in biological systems.

(Refer Slide Time: 09:35)

The Maxwell's equations in dif	ferential form are usually useful	*
To convert the integral Maxwe	Il equations into their differential forms, we used two theorems in mathematics	TE
Gauss' theorem states:	$\oint_{S} \vec{D} \cdot d\vec{A} = \int_{T} (\vec{V} \cdot \vec{D}) dV$	
	Relationship between relevant surface and volume integrals	
Stokes' theorem states:	$\oint_C \vec{D} \cdot d\vec{S} = \int (\vec{\tau} \times \vec{D}) d\vec{A}$	
	Relationship between relevant contour and surface integrals	
		-

Such as.

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If we apply Gauss' theorem to Maxwell's first equation $\oint_{\sigma} e_{\sigma} \vec{E}. d\vec{A} = \int_{V} \rho dV$	*
LHS: $\oint e_{\nu}\vec{E}.d\vec{A} = \int_{V} (\vec{V} \cdot e_{\nu}\vec{E}) dV$	NPTEL
$=\int_V \rho dV$	
Since dV is arbitrary,	
$(\vec{v} \cdot \epsilon_0 \vec{k}) = \rho$ Differential form of Maxwell's first equation	
	L'ÀU

The differential forms.

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Now, let us consider Maxwell's second equation  $\oint_{c} \vec{H} \cdot d\vec{S} = \int_{\vec{S}} \vec{I} \cdot d\vec{A} + \frac{d}{dt} \int_{\vec{S}} e_{\vec{a}} \vec{E} \cdot d\vec{A}$  $\oint_C \vec{D} \cdot d\vec{S} = \int_C (\vec{\nabla} \times \vec{D}) d\vec{A}$ From Stokes' theorem, we know Thus, the LHS of Maxwell's second equation becomes  $\int (\vec{v} \times \vec{H}) d\vec{A}$  $\int_{S} \left( \vec{V} \times \vec{H} \right) d\vec{A} = \int_{S} \vec{I} \cdot d\vec{A} + \frac{d}{dt} \int_{S} \epsilon_{0} \vec{E} \cdot d\vec{A}$ Therefore, The surface S is fixed in time. Thus, the derivative can be taken inside the integral. Also, S is arbitrary. Thus,  $(\vec{\nabla} \times \vec{H}) = \vec{I} + \frac{d}{dt} (\epsilon_0 \vec{E})$ Differential form of Maxwell's second equa

How to convert the integral forms and the differential forms and the differential forms we saw by using the 2 theorems the Gauss's theorem, the Stokes theorem in mathematics.

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On similar lines, the other two equations can also be converted to their Also, we have a differential form of the charge conservation equation, for Let us list all of them here		
$\left(\vec{\vec{v}}\cdot \epsilon_{0}\vec{\vec{k}}\right)= ho$	Eq. 5.5 - 1	
$\left(\vec{\nabla}\times\vec{H}\right) = \vec{I} + \frac{d}{dt}\left(\epsilon_{o}\vec{E}\right)$	Eq. 5.5 ~ 2	
$\left( ec{v}  imes ec{E}  ight) = - rac{d}{dt} ig( \mu_a ec{H} ig)$	Eq. 5.5 – 3	
$\left( \overline{ ec r}  \cdot  \mu_0 \overline{H}  ight) \; = \; 0$	Eq. 5.5 ~ 4	
$\frac{d\rho}{dt} + \vec{v} \cdot \vec{t'} = 0$	Eq. 5.3 - 5	-
		A-P

And then these were the differential forms, on top of that we said that these are fine for free space, for biological systems, you do not have free space, you have a medium. And therefore you cannot use the permittivity of free space.

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You will have to use the permittivity of the medium, however the permeability of free space is does not change in these equations. So, the same equations that we had earlier can be used just by changing  $\varepsilon_0$  to  $\varepsilon$ . The effects that we discussed because of the presence of the medium are polarization and magnetisation.

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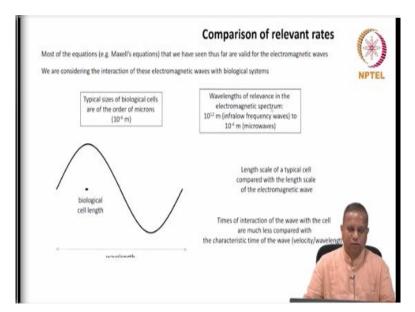
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## (Refer Slide Time: 10:37)

		G
In a medium in the presence of an electric field	d, there could be free charges and polarization charges	
Let the charge density due to free charges be	$p_{fc}$	NPTEL
Let the charge density due to polarization char	rges be $\rho_{pc}$	
The Gauss' law for this system can be written a	35	
$\vec{v} \cdot v\vec{E} = \mu$	$\rho_{fc} + \rho_{pc}$	
free space permittivity, $\epsilon_{o}$ , by the medium per	media remain the same with the replacement of the mittivity, e medium in which its properties do not change with space/pos	ition
interestingly, the permeability of most biologic approximated very well to $\mu_{\rm o}$	cal materials such as cells and tissues, can be	-

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Then we also saw the.

### (Refer Slide Time: 10:49)

In other words, The rates of interactions of the waves with the biological entities are much faster compared to the rates of variation in the wave characteristics	
Thus, the interactions of the waves with the biological entities can be considered to be at pseudo-steady state when compared to the wave processes Recall, that under PSS conditions, the variations in the rates of the much faster process can be ignored if the interest is in the slower process. Here the equations of interest (e.g. Maxwell's equations) describe the slower (wave) process	
Therefore, we can ignore the time derivatives in the relevant equations, say the Maxwell's equations These are called the electro-quasi-state (EQS) and the magneto-quasi-state (MQS) approximations	
	-

Electro-quasi state approximation and the magneto-quasi state approximation which effectively delete the time variations and the time derivatives in the Maxwell's equations.

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$(\vec{r} \cdot \epsilon \vec{E}) = \rho$	Eq. 5.5 – 6	NPTEL
$(\vec{v} \times \vec{n}) = \vec{i}$	Eq. 5.5 – 7	
$(\vec{r} \times \vec{E}) = 0$	Eq. 5.5 – 8	
$(\vec{v} \cdot \mu_o \vec{H}) = 0$	Eq. 5.5 - 9	
$(\mathbf{v} \cdot \boldsymbol{\mu}_0 \boldsymbol{\mu}) = 0$	14.33-3	-
		H
	and the second second	CONCERCION OF

And they become simpler for us to use for interactions with biological situation, ok and biological systems that is, so these are the equations of relevance. We also found that, we also saw that these equations can be used for any situation, we saw how you could use this to get a capacitor equation and the bile lipid bile layer membrane can be viewed as a capacitor.

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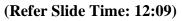
An electrical potential, V, is relate	$\vec{E} = - \vec{\nabla} \mathbf{V}$	Eq. 5.5. – 10	NPT
Therefore,	$\vec{\nabla} \cdot \epsilon \vec{\varepsilon} = \vec{\nabla} \cdot \left(-\epsilon  \vec{\nabla} \mathbf{V}\right)$		
Substituting this in Eq. 5.5 – 6,	$\vec{v} \cdot (-\epsilon \vec{v} \mathbf{V}) = \rho$		
Thus,	$\nabla^2 \mathbf{V} = -\frac{\rho}{\epsilon}$	Eq. 5.5. – 11	
	Poisson equation		
In the region where no charges a	re present ( $\rho = 0$ )		-
	$\nabla^2 \mathbf{V} = 0$	Eq. 5.5 12	
	Laplace equation		9.01
	analysis of biological systems, e.g. cert		T

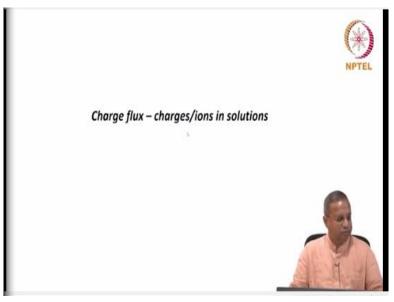
Right, then we saw the Poisson equation, Laplace equation which are useful equations for analysis.

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	Constitutive equation	A
Let us recall that Fick's law was a constitutive equation		NPTE
It related diffusive flux and concentration gradient, and is valid for a	class of materials	
For certain materials, the charge flux is proportional to the potential	l gradient	
$\vec{l}' = -k_x \vec{r} V = k_x \vec{E}$	Eq. 5.6. – 1	
$k_{\rm c}$ is the electrical conductivity of the medium (typical unit: Si	emens cm <sup>-1</sup> )	
Eq. 5.6. – 1 is a constitutive equation, which is valid for a class of ma	terials	
Ohm's law		201
		V
	NEN	4 4

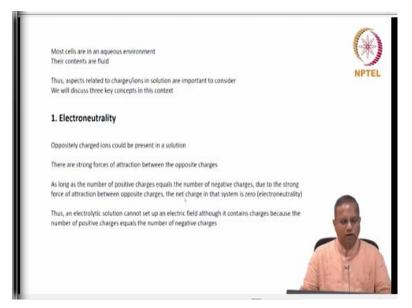
Then we saw Ohm's law as a constitutive equation, ok that is I think that is where we finished up there. And then we saw some Maxwell application of Maxwell's equation as I Just mentioned. And we also said that the fundamental equations on which EEG is based, this also derived from Maxwell's equation. For that matter anything is fundamentally derived from Maxwell's equation. Then we saw the aspects of charges and solution 3 different important aspects.





Related to charges in solution.

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Or ions in solution, first one was electroneutrality.

## (Refer Slide Time: 12:18)

Let us say a charge densit	n time tic solution (medium) which is homogenou $y(say \rho_{\phi})$ is added to the above solution utring, the charge will be conducted away		
	hat it takes for achieving the new equilibri		
According to Ohm's law, t			
	$\vec{l} = k_e \vec{E}$	Eq. 5.7.2 1	
	$\vec{l} = k_e \vec{E}$ $\vec{E} = - \vec{\nabla} \mathbf{V}$		
Note: the electrolytic solu	tion is a homogenous conductor (or k <sub>e</sub> is c	onstant)	-
	$\vec{\nabla} \cdot \vec{I} = k_x \vec{\nabla} \cdot \vec{E}$	Eq. 5.7.2 2	
	25.		- E
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			1000

The second one was charge relaxation time and the third aspect.

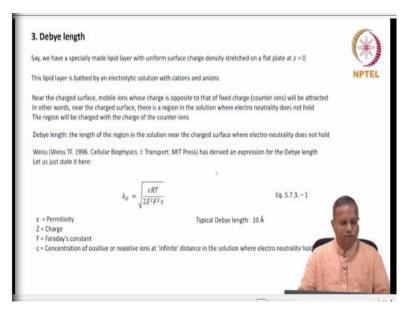
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From Gauss' law $(\vec{v} \cdot e_0 \vec{E}) = \rho$ Eq. 5.7.2. – 3 $\rho = \text{charge density at any time, t}$ Substituting 5.7.2 – 3 into 5.7.2 – 2 $\vec{v} \cdot \vec{l} = \frac{k_x}{\epsilon} \rho$ Eq. 5.7.2. – 4 If we assume that upon application of $\rho_{in}$ , the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5	$\rho = \text{charge density at any time, t}$ Substituting 5.7.2 – 3 into 5.7.2 – 2 $\vec{v} \cdot \vec{l} = \frac{k_x}{\epsilon} \rho$ Eq. 5.7.2 – 4 If we assume that upon application of $\rho_{in}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{l'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4	$\rho$ = charge density at any time, t Substituting 5.7.2 – 3 into 5.7.2 – 2 $\vec{v} \cdot \vec{l} = \frac{k_x}{\epsilon} \rho$ Eq. 5.7.2 – 4 If we assume that upon application of $\rho_{in}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5				
Substituting 5.7.2 – 3 into 5.7.2 – 2 $\vec{r} \cdot \vec{r} = \frac{k_x}{\epsilon} \rho$ Eq. 5.7.2. – 4 If we assume that upon application of $\rho_{uv}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation	Substituting 5.7.2 – 3 into 5.7.2 – 2 $\vec{v} \cdot \vec{l} = \frac{k_{\pi}}{\epsilon} \rho$ Eq. 5.7.2. – 4 If we assume that upon application of $\rho_{in}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{l'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4	Substituting 5.7.2 – 3 into 5.7.2 – 2 $\vec{v} \cdot \vec{l} = \frac{k_{\pi}}{\epsilon} \rho$ Eq. 5.7.2. – 4 If we assume that upon application of $\rho_{in}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{l'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4	From Gauss' law		Eq. 5.7.2 3	()
If we assume that upon application of $\rho_{\rm in}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation	If we assume that upon application of $\rho_{uv}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4	If we assume that upon application of $\rho_{uv}$ the increase in charge density in the solution is uniform, we can use the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4	Substituting 5.7.2 – 3	i into 5.7.2 –2		NPT
the charge conservation (charge continuity) equation	the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4	the charge conservation (charge continuity) equation $\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5 Substituting Eq. 5.5. – 5 in 5.7.2 – 4		$\vec{\nabla}\cdot\vec{I}\;=\;\frac{k_{\sigma}}{\epsilon}\;\rho$	Eq. 5.7.2. – 4	
$\vec{v} \cdot \vec{t'} = -\frac{\partial \rho}{\partial t}$ Eq. 5.5. – 5	Substituting Eq. 5.5. – 5 in 5.7.2 – 4	Substituting Eq. 5.5. – 5 in 5.7.2 – 4			in the solution is uniform, we can use	
				$\vec{v}\cdot\vec{t'}=-\frac{\partial\rho}{\partial t}$	Eq. 5.5 5	

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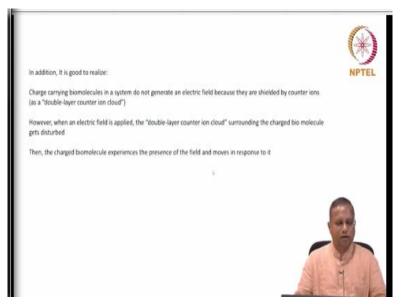
$r_r = \frac{\epsilon}{k_c}$ is the charge relaxation time For water: $k_r = 0.01 \text{ S cm}^3$ $\epsilon = 80 k_0 \approx 80 \text{ X 8.85 X 10}^{12} \text{ F cm}^3$ Then, $r_r = 0.7 \text{ ns}$ Charges can be relaxed rapidly Charge relaxation needs to be considered while designing systems in which the biological materials – cells,		${ m r}_r=rac{\epsilon}{k_e}~~{ m is~the~chs}$	rge relaxation time	NPT
$\begin{array}{l} k_{F}{=}~0.01~{\rm S~cm^{-1}}\\ e=80e_{0}{*}~80~{\rm X~8.85~X~10^{+2}~F~cm^{-1}}\\ {\rm Then}, \qquad r_{F}{=}~0.7~{\rm ns}\\ \end{array}$ Charges can be relaxed rapidly			2	
Charges can be relaxed rapidly	$k_e$ = 0.01 S cm <sup>-1</sup> $\epsilon$ = 80 $\epsilon_o$ = 80 X B.			
	Charges can be relaxed rap	sidty	which the biological materials – cells,	

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Was the Debye length, ok. The region where the electroneutrality does not hold especially when you have a surface of when you have a charge surface in solution, ok. These are all relevant aspects for analysis, I think this is what we saw.

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Till this chapter, we also saw that it is good to realize charge carrying biomolecules in a system, do not generate an electric field because they are shielded by counter ions as a double layer counter ion cloud. However when the electric field is applied the double layer counter ion cloud surrounding the charged biomolecules get disturbed. And then the charged biomolecule experiences the presence of the field and moves in response to that ok.

Ok, we will take a break here, I think it is time then when we meet for the next lecture the last review lecture you would look at multiple driving forces being responsible for a flux or multiple fluxes, ok. Predominantly a flux that is the review of the last chapter, see you then.